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# Activatable organic photoacoustic probes for *in vivo* anion imaging

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The merits of non-invasiveness, non-ionization and impressive tissue penetration depth of photoacoustic (PA) imaging technology make it attractive for *in vivo* detection. Analyte-responsive organic PA probe-based *in vivo* imaging of bio-functional small anions shows advances in real-time visualization of biological action, early diagnosis and precision therapy of diseases, due to the significance of such anion species in various physiological and pathological processes. In this mini review, the anion-recognition mechanisms of PA probes are particularly focused, from the perspective of chemistry, to exemplify the structural design strategies of activatable organic PA probes. Finally, the challenges and outlooks are discussed to pave the way for further development of PA probes for *in vivo* imaging of bio-functional small anions, which provides an alternative approach to interpret anion-mediated biological events.

## 1. Introduction

Bio-functional small anions have attracted broad attention recently due to their important roles in various physiological and pathological processes. For instance, fluoride ions ( $F^-$ ) are closely related to osteoporosis and some dental diseases.<sup>1–3</sup>

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Reactive oxygen, nitrogen, and sulfur anion species (RONs and RSSs), such as superoxide anions ( $O_2^{\bullet-}$ ), hypochlorite ( $HClO/CLO^-$ ), peroxyxynitrite ( $ONOO^-$ ), and thiolates ( $RS^-$ ), show significance in redox homeostasis and signal transduction in the living body.<sup>4</sup> Their overproduction results in oxidative or nitrosative stress, which disrupts intracellular redox homeostasis and further affects gene transcription, cellular signaling, the activity of various enzymes and bio-macromolecules, and the functionality of cells and organs. Oxidative stress could lead to free radical overaccumulation and further produce oxidative damage by lip peroxidation and oxidative modification of proteins.<sup>5</sup> Prolonged cellular or tissue damage may induce



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severe diseases such as cancers and cardiovascular diseases.<sup>6</sup> Therefore, cellular redox homeostasis is indispensable for maintaining the functioning of bio-macromolecules and cells. In this context, it is worthwhile to specifically detect these bio-functional anion species *in vivo*, which could afford plenty of vital information for the evaluation of the redox status of tissues.<sup>7</sup> Moreover, an *in vivo* anion imaging approach could be utilized for imaging of oxidative tissue injury or tumor imaging.<sup>8,9</sup> *In vivo* imaging of these anion species could make a complementarity with *in vitro* detection to elucidate the bio-functions of anions and acquire a comprehensive understanding of biological events, assisting in the theranostics and prognosis of diseases.<sup>10</sup>

To date, several optical<sup>11–13</sup>/non-optical<sup>14,15</sup> detection technologies, *e.g.*, positron emission tomography (PET), computed tomography (CT), ultrasound imaging (US), magnetic resonance imaging (MRI), and fluorescence imaging (FI) have been widely used for *in vivo* imaging. However, PET and CT are based on ionizing radiation, which increases the risk of radiation damage. US affords deep tissue penetration but is subject to low imaging resolution and contrast. MRI modality is much more expensive and unavailable for implant-carrying patients. The popular FI features excellent sensitivity and specificity, gratifying resolution and biological safety yet still suffers from serious light scattering of tissues, which causes undesirable signal interference and consequent low contrast and shallow imaging depth. By contrast, emerging photoacoustic imaging (PAI) is a promising technology, which collects ultrasound signals after light excitation. It combines optical and ultrasound technologies to achieve deep tissue penetration and, meanwhile, a satisfactory spatiotemporal resolution for *in vivo* imaging.<sup>16,17</sup> Furthermore, PAI could cooperate with other detection technologies to realize multimodal imaging as well as obtain qualitative or quantitative information on varying scales in a more accurate approach.<sup>18–20</sup> On the basis of this

background, much effort has been devoted to the development of PA contrast agents, especially analyte-specifically responsive PA probes.<sup>21,22</sup> This kind of activatable PA probe displays signal variations after interaction with a specific analyte, showing the merits of excellent selectivity and a higher signal-to-noise ratio (S/N) than non-responsive PA contrast agents, due to the low background interference.<sup>23,24</sup> According to the varying signal outputs, this type of activatable PA probe could be divided into turn-on, turn-off and ratiometric types. After interacting with analytes, turn-on/off probes exhibit an intensity increase/decrease at the wavelength of the maximum PA signal. The turn-on/off times are used to appraise the amplitude of the signal changes. By contrast, ratiometric probes show signal peak shifts, by which the background interference is diminished instinctively.<sup>25</sup> Besides the single-stimulus activatable PA probe, dual- and even multi-stimuli responsive probes are attractive due to the less immature response interference.<sup>26</sup> An ideal activatable PA probe should have high absorptivity (molar extinction coefficient  $> 10^4 \text{ M}^{-1} \text{ cm}^{-1}$ ) in the first and second windows of the near-infrared (NIR-I and II) region avoiding the interruption of tissue and water, as well as possess a low quantum yield, excellent photothermal conversion ability and distinguished biological characteristics.<sup>23,27,28</sup> It seems a feasible shortcut to design PA probes from the existing NIR fluorescent (NIRF) probes, and there are indeed many successful examples. In the meantime, more factors should be carefully weighed up in a balanced manner. For instance, the fluorescence quantum yield, theoretically, should be low to intensify the nonradiative energy dissipation and then generate an amplified PA signal. Whereas a long-lived first excited state ( $S_1$ ) with a high quantum yield conversely facilitates PA generation. Besides, the rotatable bonds within the probe could also improve PA emission by promoting nonradiative energy dissipation. For example, there is not any measurable PA signal detected from the classic fluorescent dyes methylene blue and



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Rhodamine800 due to the lack of rotatable bonds.<sup>29,30</sup> Other parameters including excited state absorption, triplet state contributions, relaxation kinetics, solubility, and photo-bleaching<sup>31–33</sup> likewise affect the PA signal amplitude. Consequently, it remains challenging to design a well-performed PA probe.

The fundamental photophysical characteristics and typical structure of PA contrast agents can be found elsewhere<sup>28,30,34</sup> and will not be discussed in detail herein. The categories and various applications of PA probes have also been reviewed a lot.<sup>17,31,35–38</sup> Recently, Pu<sup>39</sup> and Chan *et al.*<sup>40</sup> gave overviews individually, focusing on PA probes for either different tissue imaging or tumor microenvironment profiling, highlighting the assembling methods and the performance of the nanoprobes. Tang *et al.* discussed the assembly strategies of organic dyes for FI and PAI.<sup>41</sup> However, there is no review article yet, to the best of our knowledge, systematically summarizing the recent development of activatable organic probes for *in vivo* anion imaging, and focusing more on the specific anion-recognition mechanisms of the probes at the molecular level. On the other hand, as aforementioned, *in vivo* anion imaging is of significance to early diagnosis and therapy of various diseases, and PAI shows unique superiority for *in vivo* imaging, which together causes an increasing number of original research papers published on the topic. Herein, we review the state-of-the-art studies to highlight the interactions between bio-functional anion species and activatable organic PA probes from the perspective of chemistry. The involved imaging mechanisms include cleavage, ring-opening, oxidation, degradation, substitution, and reduction reactions and the deprotonation process (Fig. 1). This review aims to imply future perspectives of this field to promote the extensive development of this kind of scarce PA probe.

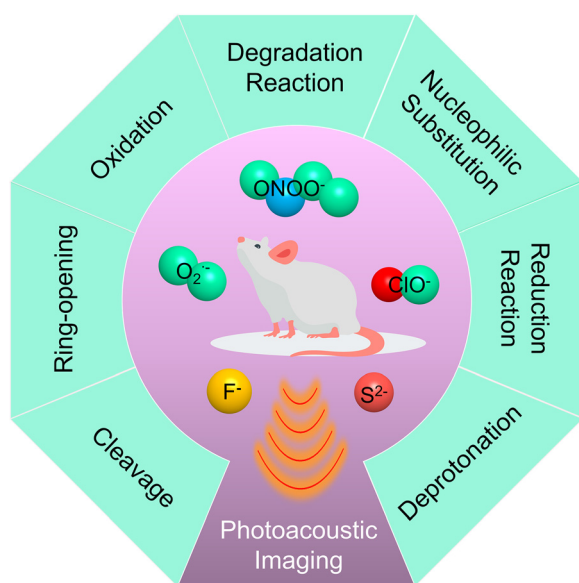


Fig. 1 An overview of activatable organic PA probes for *in vivo* anion imaging, with an emphasis on activation mechanisms.

## 2. Activatable organic PA probes for *in vivo* anion imaging

Essentially, the anion-recognition process mainly involves the specific analyte-induced chemical structural variation of the probe, further resulting in the change of their photo-chemical and -physical properties (*e.g.*, the aggregation state, charge transfer pathway), followed by the switch of the PA signal. In this section, the specific recognition mechanisms of activatable PA probes to bio-functional anion species are elaborately exemplified to clarify the probe design strategies.

### 2.1 *In vivo* imaging of reactive oxygen and nitrogen anion species

RONSs and RSSs are significant for redox homeostasis in all living systems. RONSs are highly active bio-oxidants. Among them, ionic RONSs  $O_2^{\bullet-}$ ,  $ONOO^-$ , and  $HClO/ClO^-$  are moderately unstable, with lifetimes of milliseconds or longer, and amenable to measurement. Moreover, they also show higher hydrophilicity than the lipophilic RONS-like singlet oxygen ( $^1O_2$ ) and hydrogen peroxide ( $H_2O_2$ ). Therefore, rational nano-engineering of PA probes could improve the selectivity towards these longer-lived polar RONSs.<sup>42</sup> Several activatable organic PA probes for imaging of these ionic RONSs are summarized in Table 1. The mechanisms include (i) cleavage; (ii) intramolecular spirolactone ring-opening; (iii) oxidation; and (iv) degradation reaction of PA probes towards ionic RONSs.

#### 2.1.1. Cleavage

**Ratiometric probes.** It is believed that  $O_2^{\bullet-}$  is the precursor of many other ROSSs. This reactive species can lead to the cleavage of specific covalent bonds to induce structural changes. In this context, an  $O_2^{\bullet-}$ -responsive semiconductor polymer nanoprobe (RSPN) was developed by Zhang *et al.* It consisted of  $O_2^{\bullet-}$ -responsive molecules (ORMs),  $O_2^{\bullet-}$ -inert semiconducting polymers (OIMs), and amphiphilic polymers (DSPE-PEG<sub>2000</sub>).<sup>43</sup> After being attacked by  $O_2^{\bullet-}$ , the trifluoromethanesulphonyl group in the ORM skeleton was cleaved, and then the ORM could offer a robust “turn on” signal at around 690 nm, while a negligible change was observed at 800 nm which could be attributed to OIMs (Fig. 2a and b). Notably, the PA signal showed the same variation trend with absorption spectra (Fig. 2c), and the ratio of  $PA_{690}/PA_{800}$  exhibited a good linear relationship with the concentration of  $O_2^{\bullet-}$ . With respect to the selectivity, RSPN enables the specific differentiation of  $O_2^{\bullet-}$  from other common reactive species such as  $H_2O_2$ ,  $\cdot OH$ ,  $NO_2^-$ ,  $NO$ ,  $ONOO^-$ ,  $HClO$ ,  $Cys$ ,  $Hcy$ , and  $S^{2-}$ . Therefore, the ratiometric PAI ( $PA_{690}/PA_{800}$ ) endowed RSPN with reliable visualization of  $O_2^{\bullet-}$ , which provided a noninvasive and high target-to-background ratio tool for examination of atherosclerotic plaque, in particular, for atherosclerotic complicated with pneumonia (Fig. 2d).

As a rapid reaction product of  $O_2^{\bullet-}$  and  $NO$ ,  $ONOO^-$  also has strong oxidizing ability to break chemical bonds. For instance, Pu's group reported two PA probes to settle scientific problems of  $ONOO^-$  sensing. One of them is the bulky borane-doped organic semiconducting nanoprobes (OSN-B1) constructed *via*



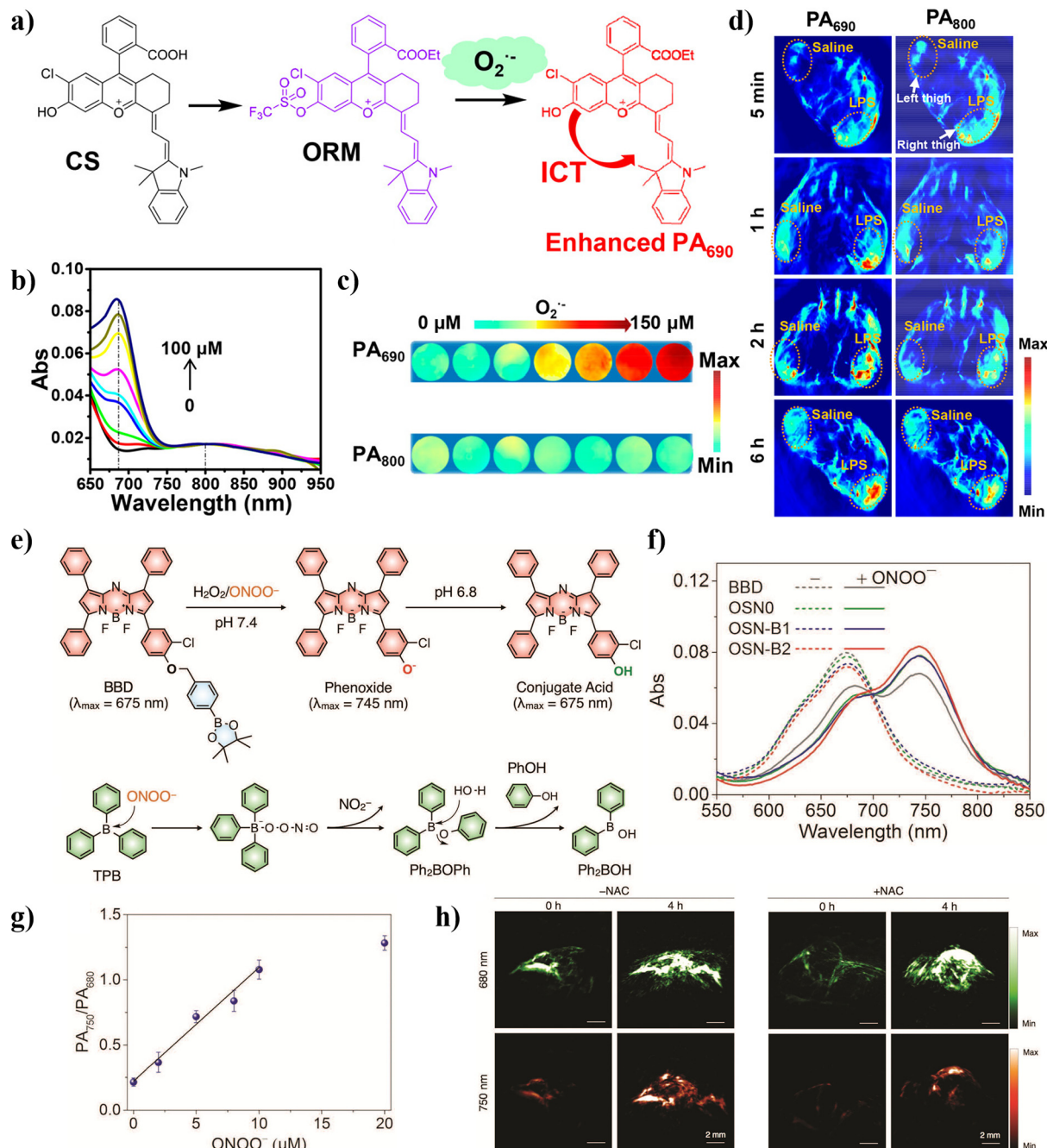
Table 1 Summary of the representative activatable organic PA probes for *in vivo* imaging of anions

Probe	Ion species	Reaction site	Mechanism	PA signal	Times turn-on ( <i>in vivo</i> )	LOD	Animal model	Dose ( $\times 10^{-3}$ mg)/injection method	Cytotoxicity (MTT assay)	Ref.
RSPN <sup>db</sup>	O <sub>2</sub> <sup>•-</sup>		Cleavage	PA <sub>690</sub> / PA <sub>800</sub>	—	—	Plaque-bearing mice (with or without pneumonia)	14 <i>i.v.</i>	40 $\mu\text{g mL}^{-1}$ (ORM)	43
OSNs <sup>db</sup>	ONOO <sup>-</sup>		Cleavage	PA <sub>745</sub> / PA <sub>675</sub>	2.6-fold	100 nM	4T1 xenograft tumor	25 <i>i.v.</i>	100 $\mu\text{g mL}^{-1}$	44
CySO <sub>3</sub> CF <sup>ea</sup>	ONOO <sup>-</sup>		Cleavage	PA <sub>680</sub>	2.1-fold	145 nM	4T1 xenograft tumor	3.8 <i>i.v.</i>	80 $\mu\text{mol L}^{-1}$	45
PA-MMSiNQ <sup>da</sup>	HOCl		Spirolactone ring-opening	PA <sub>660</sub>	—	—	Mouse thigh	2.9 <i>s.c.</i>	—	46
SiRho-HD <sup>ec</sup>	ONOO <sup>-</sup>		Oxidation	PA <sub>715</sub>	—	1.3 $\mu\text{M}$	Acute kidney injury	10.5 <i>i.v.</i>	10 $\mu\text{M}$	47
BDP-DOH <sup>db</sup>	O <sub>2</sub> <sup>•-</sup>		Oxidation	PA <sub>750</sub> / PA <sub>680</sub>	6.3-fold	30 nM	EMT6 xenograft tumor	10 <i>i.v.</i>	100 $\mu\text{g mL}^{-1}$	48
ONc/IR780 @F127 <sup>db</sup>	ONOO <sup>-</sup>		Degradation	PA <sub>860</sub> / PA <sub>775</sub>	2.34–2.22-fold (tumor vs. thigh)	10 nM	4T1 xenograft tumor and thigh	25 (ONc) <i>i.v.</i>	10 $\mu\text{g mL}^{-1}$ (ONC)	49
SOA <sup>db</sup>	ClO <sup>-</sup>		Degradation	PA <sub>780</sub> / PA <sub>680</sub>	1.47-fold (6 h vs. 2 h post-injection)	700 nM	4T1 xenograft tumor	50 <i>i.v.</i>	50 $\mu\text{g mL}^{-1}$	50
NIR-S <sup>eb</sup>	Cys/ Hcy		Nucleophilic substitution	PA <sub>695</sub> / PA <sub>840</sub>	> 10-fold	—	Leg	4.6 <i>i.v.</i>	50 $\mu\text{M}$	54
ZNNPs <sup>eb</sup>	H <sub>2</sub> S		Nucleophilic substitution	PA <sub>680</sub> / PA <sub>900</sub>	2.8-fold	0.68 $\mu\text{M}$	Acute hepatotoxicity, cerebral hemorrhage, HCT116 xenograft tumor	10 mg kg <sup>-1</sup> <i>i.v.</i>	100 $\mu\text{g mL}^{-1}$	57
AzHD-LP <sup>db</sup>	H <sub>2</sub> S		Reduction	PA <sub>700</sub> / PA <sub>532</sub>	—	91 nM	HCT116 xenograft tumor	1 mg kg <sup>-1</sup> <i>i.v.</i>	100 $\mu\text{g mL}^{-1}$	58
LET-1 <sup>da</sup>	F <sup>-</sup>		Deprotonation	PA <sub>680</sub>	4-fold	0.2 mM	Mouse liver	40 $\mu\text{M}$ , <i>s.c.</i> / 5 mg kg <sup>-1</sup> , <i>i.v.</i>	—	63

Note: The response type: <sup>a</sup> Turn on. <sup>b</sup> Ratiometric. <sup>c</sup> Turn off. <sup>d</sup> PA imaging. <sup>e</sup> PA/FL dual-model imaging. Fold turn-on: PA intensity enhancement of the probe-treated group *versus* control group (unless otherwise specified). LOD: the limit of detection. Model: mouse. Injection methods: *i.v.*: intravenous injection; *s.c.*: subcutaneous injection; and *i.c.*: co-incubation.

nanoprecipitation *via* the assistant of an amphiphilic polymer, PEG-*b*-PPG-*b*-PEG for ratiometric PAI of ONOO<sup>-</sup> in real-time

(Fig. 2e).<sup>44</sup> Both nanoprobes consist of a boronate-caged boron-dipyrromethene dye (BBD) as the primary ONOO<sup>-</sup>-responsive



**Fig. 2** (a) Chemical structure of ORMs and the responsive mechanism of ORMs to  $O_2^{\bullet-}$ ; (b) absorption spectra and (c) PA images at 690 or 800 nm of RSPN upon addition of  $O_2^{\bullet-}$  in PBS; (d) PA images of the LPS-stimulated acute inflammation mice model; (e) chemical structure and reaction mechanisms of BBD and bulky borane (TPB); (f) absorption spectra of OSNs and BBD after addition of  $ONOO^-$ ; (g)  $PA_{750}/PA_{680}$  of OSN-B1 as a function of the concentration of  $ONOO^-$ ; (h) *in vivo* ratiometric PAI of  $ONOO^-$ . (a–d) Reprinted (adapted) with permission from ref. 43, Copyright 2021, American Chemical Society; (e–h) reprinted (adapted) with permission from ref. 44, Copyright 2016, Wiley-VCH Verlag GmbH & Co. KGaA.

molecule, which could undergo the breakage of an ether bond to recover the intramolecular charge transfer process. To get around the unspecific response of the BBD toward both  $ONOO^-$  and  $H_2O_2$ , an  $ONOO^-$ -degradable but  $H_2O_2$ -inert shield triphenylborane (TPB) was ingeniously doped to prevent the BBD from reacting with  $H_2O_2$ . Initially, the absorption spectra of the probe showed a maximum absorption peak at 675 nm.

With the addition of  $ONOO^-$ , the absorption at 675 nm was reduced gradually, accompanied by the emergence of a new peak at 745 nm (Fig. 2f). At the saturation point, the absorbance ratio ( $Ab_{745}/Ab_{675}$ ) of the OSNs was  $\sim 1.7$ , which was 8-fold higher than that of the initial state (0.21) (Fig. 2g). The activation of probes could be explained by  $ONOO^-$ , which caused rapid oxidative cleavage of the borate ester moiety of the BBD.

The formed anionic phenoxide product showed a characteristic absorption peak at 745 nm. Furthermore, there was no absorption change of the probes in response to other ROSSs, indicating their high selectivity in the presence of  $\text{ONOO}^-$ . Finally, the nanoprobe was successfully applied for ratiometric PAI of  $\text{ONOO}^-$  in 4T1 xenograft tumor-bearing mice (Fig. 2h).

**Turn-on probes.** The co-assembly of two components often suffers from complicated synthesis procedures and the risk of pre-leakage. Later, the same research group explored a molecular probe ( $\text{CySO}_3\text{CF}_3$ ) with good water solubility, used for *in vivo* NIRF/PA dual-modal imaging of  $\text{ONOO}^-$ . It contains a trifluoromethyl ketone moiety (TFK), an  $\text{ONOO}^-$ -sensitive unit, and a caged hemicyanine dye with a zwitterionic structure (Fig. 3a).<sup>45</sup> It could only be activated by  $\text{ONOO}^-$  among various RONSs such as  $\text{H}_2\text{O}_2$ ,  $\text{OH}^\cdot$ ,  $^1\text{O}_2$ , and  $\text{ClO}^-$ . In the presence of  $\text{ONOO}^-$  and  $\text{CySO}_3\text{CF}_3$  was cleaved into the uncaged structure ( $\text{CySO}_3\text{OH}$ ), undergoing a series of cascade oxidation-elimination reactions, accompanied by changes of absorption spectra (Fig. 3b) as well as the emergence of the PA signal. Then 4T1 tumor-bearing mice were systemically administered with  $\text{CySO}_3\text{CF}_3$  for PAI of  $\text{ONOO}^-$  and the PA intensity in the tumor region was 2.1-fold higher than that of the control (Fig. 3c).

### 2.1.2. Ring-opening

**Turn-on probes.** By virtue of the ROS-induced spirocycle opening of rhodamine, Urano's group developed a ROS-responsive PA probe, namely, PA-MMSiNQ (silicon-rhodamine based NIR scaffold) by the aforementioned spirocyclization strategy for *in vivo* detection of HOCl.<sup>46</sup> The "closed" spiro-lactone probe PA-MMSiNQ is transparent in the NIR region, yet a significant absorption at 660 nm was observed in the presence of HOCl ( $\text{S/N} > 70$ ), due to the ring-opening reaction following the oxidation of the sulfur atom (Fig. 4a and b). PA-MMSiNQ also showed high selectivity to HOCl against other ROSSs.

Finally, 3D PAI of HOCl in mouse subcutis was successfully realized by the activatable PA probe (Fig. 4c).

### 2.1.3. Oxidation

**Turn-off probes.** Owing to the strong oxidizing ability of RONSs, electron donors such as O, S, N, Se, Te, and olefin groups within aromatic systems get easily oxidized as carbonyl derivatives by such reactive anions. Based on this result, Zhang's group built a ratiometric NIRF/PA dual-modal imaging probe SiRho-HD to monitor the fluctuation of  $\text{ONOO}^-$  in cisplatin-induced acute kidney injury.<sup>47</sup> SiRho-HD contains HD and Si-Rho two units with absorption peaks at 660 and 719 nm, respectively. With the addition of  $\text{ONOO}^-$ , the absorption of SiRho-HD at 719 nm was gradually reduced until it disappeared due to the oxidative destruction of the xantheno scaffold. Similarly, the PA signal at 715 nm exhibited a 3.9-fold decay at the saturation point with the addition of  $\text{ONOO}^-$ . Although negligible PA signal fluctuation was observed from most RONSs, HOCl showed some interference on the recognition of  $\text{ONOO}^-$  under acidic conditions.

**Ratiometric probes.** Based on a similar mechanism, Zhang's group synthesized a reversibly activatable PA molecule, BDP-DOH, which consisted of an NIR boron dipyrromethene (BODIPY) dye, to evaluate the *in situ* redox status of tissues *in vivo* by tracing the redox circulation of  $\text{O}_2^{\cdot-}$  and GSH (Fig. 5a).<sup>48</sup> When  $\text{O}_2^{\cdot-}$  was encountered, the *ortho*-phenolic hydroxyl unit on BODIPY could be easily and specifically oxidized, and consequently the probe absorption at 680 nm was diminished while that at 750 nm was conversely enhanced (Fig. 5b), leading to an increased ratio of  $\text{PA}_{750}/\text{PA}_{680}$  (Fig. 5c). When  $\text{O}_2^{\cdot-}$  was added, a 9.84-fold enhancement of the  $\text{PA}_{750}/\text{PA}_{680}$  ratio of BDP-DOH was recorded. However, little variation of the ratio was observed after the addition of other RONSs. Then, a nanoprobe BDP-DOH/LP was assembled with BDP-DOH and an amphiphilic lipid (LP), DSPE-PEG<sub>2000</sub>-COOH, to

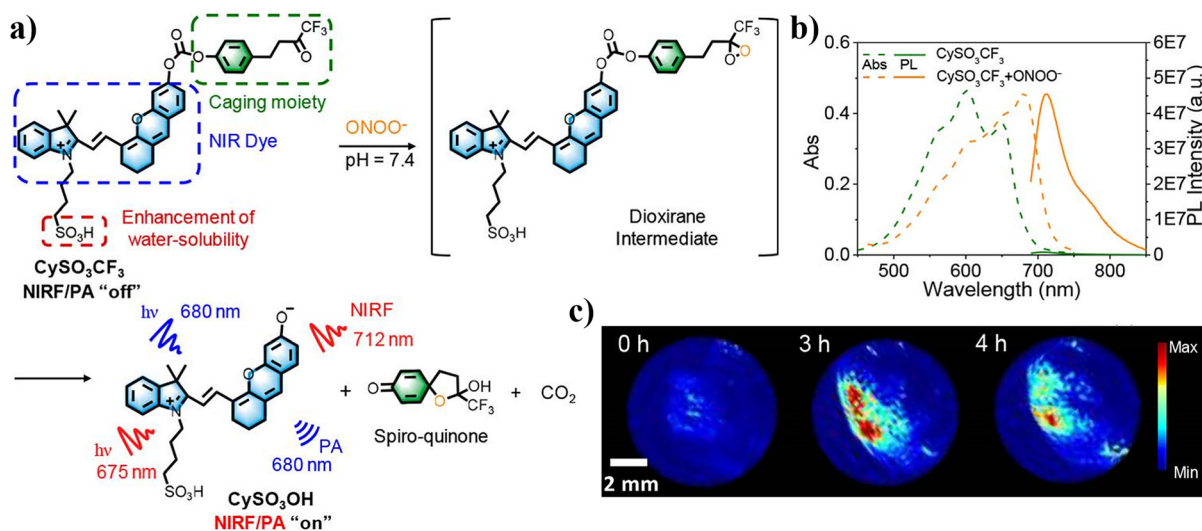


Fig. 3 (a) Chemical structure and sensing mechanism of  $\text{CySO}_3\text{CF}_3$  to  $\text{ONOO}^-$ ; (b) UV-vis absorption (dashed line) and fluorescence (solid line) spectra of  $\text{CySO}_3\text{CF}_3$  in the absence or presence of  $\text{ONOO}^-$ ; (c) representative PA imaging projection of tumors from a systemic administration of a living mouse at 0, 3, and 4 h post-injection of  $\text{CySO}_3\text{CF}_3$ . Reprinted (adapted) with permission from ref. 45, Copyright 2018, American Chemical Society.



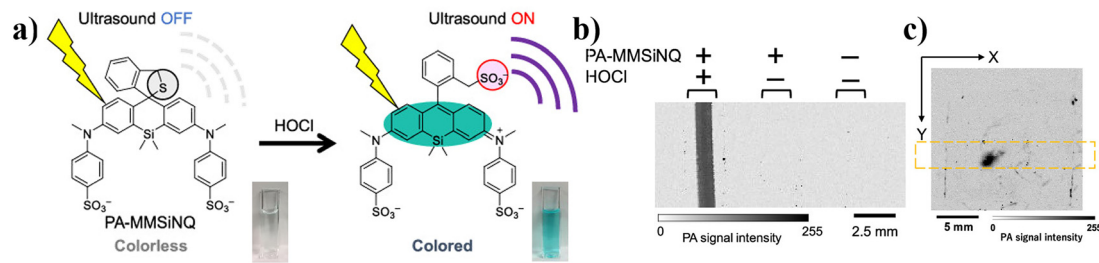


Fig. 4 (a) Chemical structure and sensing mechanism of PA-MMSiNQ for HOCl; (b) PA image of PA-MMSiNQ after (left) and before (middle) reaction with HOCl in PBS and PBS only (right); (c) PA images in living mice. Reprinted (adapted) with permission from ref. 46, Copyright 2019, American Chemical Society.

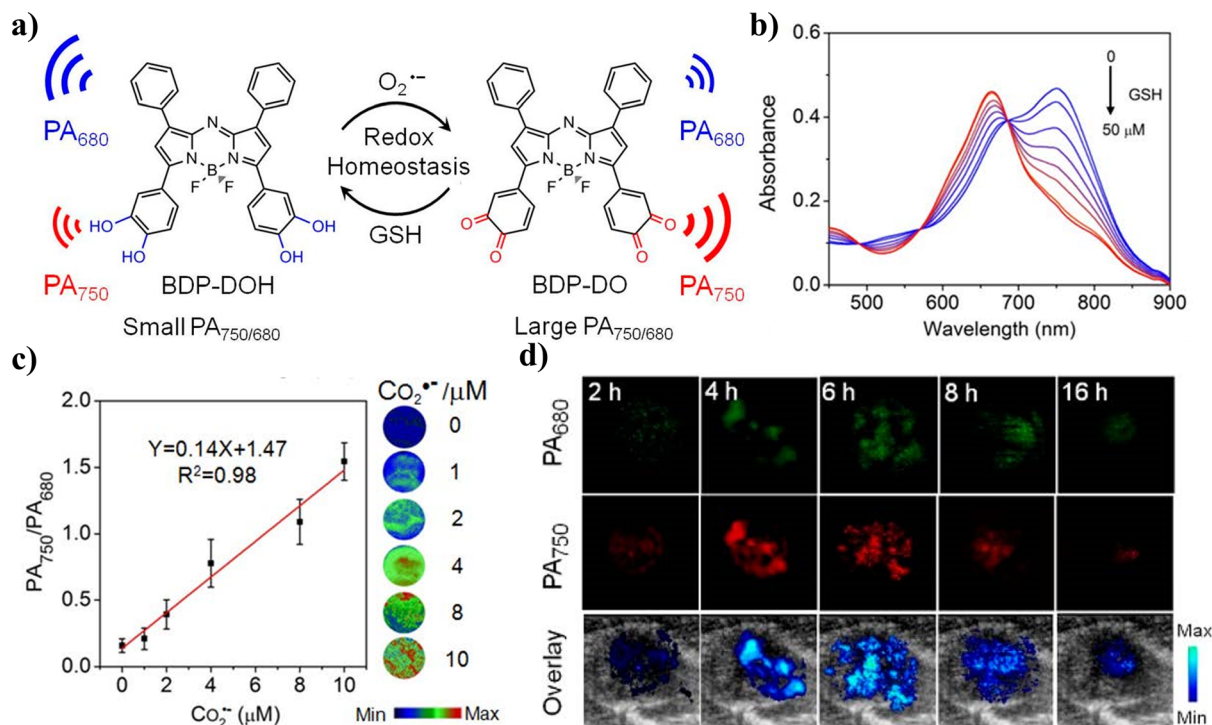


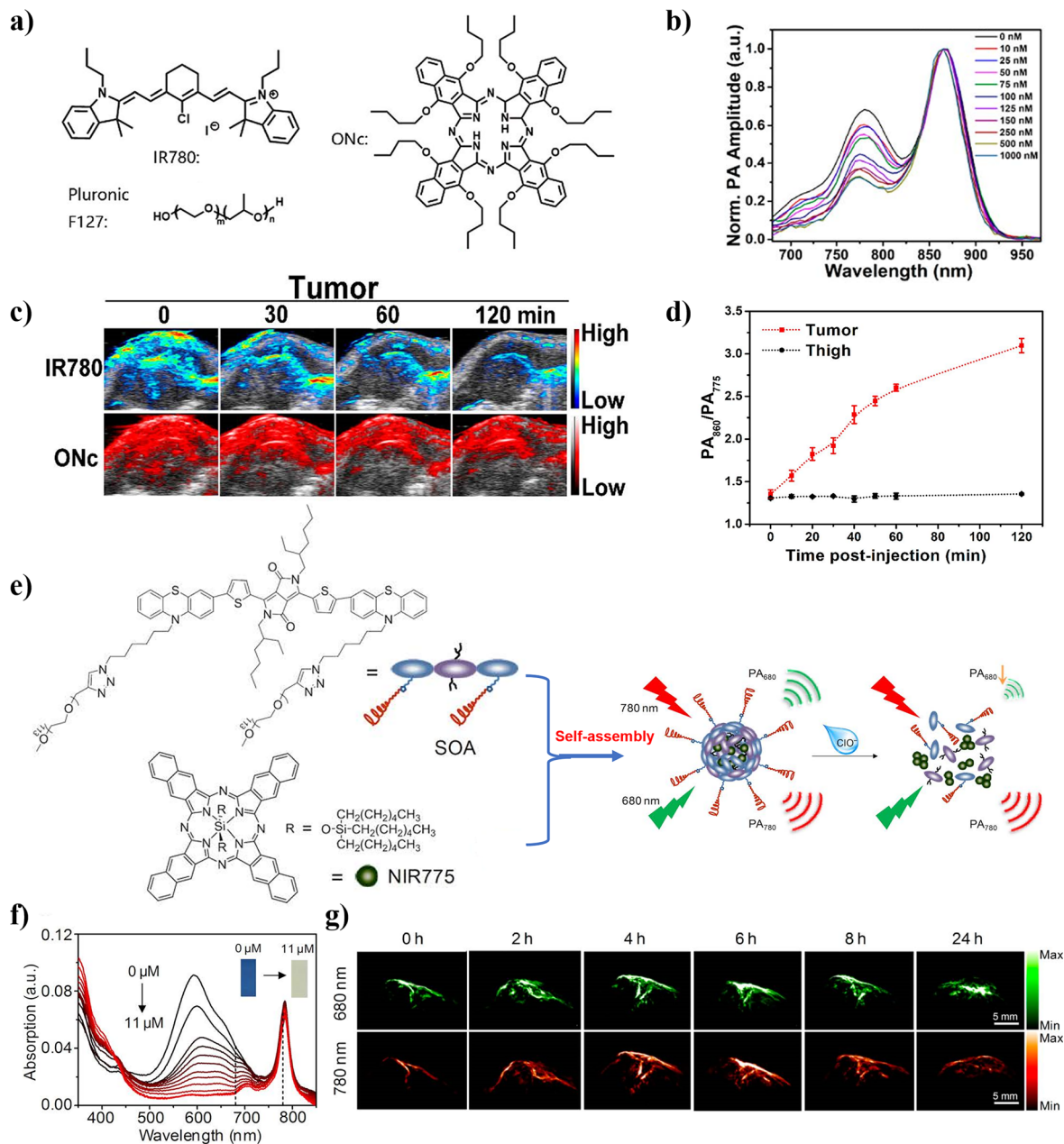
Fig. 5 (a) Chemical structure and the reversible PA sensing mechanism of BDP-DOH towards  $O_2^{\bullet-}$  and GSH; (b) absorbance changes of BDP-DOH upon titration with  $O_2^{\bullet-}$ ; (c) the  $PA_{750}/PA_{680}$  intensity of BDP-DOH with different concentrations of  $O_2^{\bullet-}$ ; (d) real-time PA and US (grey) images of the redox status in the tumor-bearing mouse pre-treated with PMA. Reprinted (adapted) with permission from ref. 48, Copyright 2019, American Chemical Society.

optimize the biological characteristics of BDP-DOH and to facilitate the delivery into tumor tissues *via* the improved permeability and retention effect (Fig. 5d).

#### 2.1.4. Degradation

**Ratiometric probes.** Degradation of PA probes usually leads to signal attenuation, which thereby displays an undesirable turn-off PA response to analytes. To overcome the obstacle, an inert reference compound can be co-assembled with the responsive probe *via* nano-engineering to achieve the ratiometric PA signals towards specific target analytes. Naphthalocyanine is a commonly used NIR reference dye due to its excellent stability towards RONSs. Lin's group fabricated such a ratiometric PA nanoprobes which contained an  $ONOO^-$ -sensitive IR780 iodide

dye and an  $ONOO^-$ -inert internal standard, 5,9,14,18,23,27,32,36-octabutoxy-2,3-naphthalocyanine (ONC) (Fig. 6a).<sup>49</sup> With the addition of  $ONOO^-$  into the probe solution, the PA signal at 775 nm declined due to the  $ONOO^-$ -induced oxidative degradation of IR780 (Fig. 6b), while that of ONC at 860 nm remained nearly unchanged. Moreover, the  $PA_{860}/PA_{775}$  ratio of the probe was  $\sim 2.88$  in response to  $ONOO^-$ , while they were less than 1.5 for other RONS analogues and RSSs ( $NO^-$ ,  $ClO^-$ ,  $O_2^{\bullet-}$ ,  $H_2O_2$ ,  $ROO\cdot$ ,  $\cdot OH$ ,  $^1O_2$ , Cys, Arg), exhibiting high selectivity. After intratumoral injection of the nanonaps, the  $PA_{860}/PA_{775}$  ratio of tumor tissues gradually rose up to  $3.10 \pm 0.084$  at 120 min-post-injection, while that of the control group remained almost unvaried ( $1.36 \pm 0.001$ ) within 120 min



**Fig. 6** (a) Chemical structures of IR780, pluronic F127, and ONc used for the preparation of nanonaps; (b) PA spectra of nanonaps with different concentrations of ONc; (c) *in vivo* ratiometric PAI performance of the nanonaps to ONOO<sup>-</sup> in 4T1 tumor tissues; (d) PA<sub>860</sub>/PA<sub>775</sub> ratios as a function of post-injection time; the chemical structures of the SOA and NIR775 used for the synthesis of the PA nanoprobe *via* a self-assembly process and illustration of the sensing mechanism for the nanoprobe; (f) UV-vis absorption spectra of the nanoprobe upon addition of ClO<sup>-</sup>; (g) representative PA image of a subcutaneous 4T1 tumor in a nude mouse. (a–d) Reprinted (adapted) with permission from ref. 49, Copyright 2018, American Chemical Society; (e–g) Reprinted (adapted) with permission from ref. 50, Copyright 2017, American Chemical Society.

post-injection, achieving the ratiometric imaging of ONOO<sup>-</sup> *in vivo* (Fig. 6c and d).

Similarly, Pu and co-workers utilized stimulus-responsive semiconducting polymers to construct diverse analyte-responsive PA probes for the ratiometric detection of biomarkers *in vivo*. Among them, a PA nano-platform for the *in vivo* detection of ClO<sup>-</sup> was constructed with a semiconducting polymer (SOA) as the responsive component and the NIR775 dye as the inert reference (Fig. 6e).<sup>50</sup> In the presence of ClO<sup>-</sup>,

the characteristic absorption of SOA at 596 nm attenuated gradually due to the ClO<sup>-</sup>-caused oxidative degradation of the probe, while the absorbance at 784 nm (assigned to NIR775) remained constant (Fig. 6f). The spectral variation of the nano-platform afforded the quantitative analysis of ClO<sup>-</sup> based on the comparative analysis of the ratio of the absorbances at 780 and 680 nm (Ab<sub>780</sub>/Ab<sub>680</sub>). In good agreement with the absorption spectra, the PA<sub>780</sub>/PA<sub>680</sub> value of the nanoprobe in response to ClO<sup>-</sup> was 8.3-fold higher than that in the initial

inactivated state. After systemic administration of the nano-probe in tumor-bearing mice, the ratiometric PA signals ( $\Delta PA_{780}/\Delta PA_{680}$ ) in tumors were increasing over time and peaked at 6 h post-injection, which was  $\sim 1.47$ -fold stronger than that at 2 h post-injection, thus verifying the capability of the probe as a highly sensitive PA indicator of  $\text{ClO}^-$  in the tumors of a living mouse (Fig. 6g).

In summary, the recognition of ionic RONSs is primarily based on the oxidation reaction of the probes, though the presented final structural transformation varies. For instance, probes containing an aromatic sulphonyl group, borate ester, or carbonic ester are prone to be cleaved to form phenol anion analogues by oxidation. The ring-closed rhodamine precursor is a typical response group of RONSs *via* a ring-opening mechanism. Moreover, the *para*-position of O within xanthenes and the *ortho*-phenolic hydroxyl group are apt to be oxidized as carbonyl groups, while the cyanine skeleton is directly destroyed due to the oxidation of the polyene segment. In this context, the probe selectivity towards various RONSs may get improved by carefully considering the difference of these reactive species with respect to their oxidizing capability and then embedding the corresponding specific response moieties into the PA probes. Furthermore, nano-engineering of the probe to achieve a ratiometric PA signal may improve the S/N ratio and selectivity. For instance, by co-assembling with an inert reference, the formed ratiometric probe ONc/IR780@F127 displayed a low LOD of 10 nM toward  $\text{ONOO}^-$  with good selectivity.

## 2.2. *In vivo* imaging of reactive bio-thiol species

In contrast to RONSs, RSSs usually function as bio-reductants. Moreover, there exists a wide variety of RSSs, such as thiols (GSH, Cys, homocysteine (Hcy)),  $\text{SO}_2$ ,  $\text{H}_2\text{S}$ , and so on. The sulfhydryl anions ( $\text{RS}^-$ ) are practically reactive,<sup>51</sup> therefore, thiolates are covered in this mini review, which focuses broadly on bio-functional ions. Several representative examples of PA probes for  $\text{RS}^-$  imaging are selected to illuminate the recognition mechanisms, which mainly include reduction and nucleophilic substitution reactions. These examples might inspire potential progress in the future.

### 2.2.1. Nucleophilic substitution reaction

**Ratiometric probes.** Among various RSSs, GSH, Cys, and Hcy have piqued researchers' significant interest as they are the most important bio-thiols. Significant research efforts have been devoted to detecting them.<sup>52</sup> For instance, specific responses to thiols can be realized through the interchange of the  $-\text{S}-\text{S}-$  and  $-\text{SH}$  groups within probes.<sup>53</sup> Therefore, disulfide motifs occupy an important position in thiol detection, and drug delivery and release systems. On the other hand, as the sulfhydryl anion ( $\text{RS}^-$ ) is the practically reacting constituent of RSSs, its nucleophilic nature could be utilized for the activation of probes. For example, He's group reported a *para*-amino phenyl thioether-substituted cyanine dye probe NIR-S for NIRF and ratiometric PA dual-modal *in vivo* imaging of Cys and Hcy.<sup>54</sup> Basically, the aryl-thioether of NIR-S was subjected to nucleophilic attack by Cys/Hcy and it subsequently underwent the Smiles rearrangement

reaction, leading to a turn-on NIR fluorescence and ratiometric PA response. With the addition of Cys/Hcy, the ratio of PA intensities at 695 nm ( $\text{PA}_{695}$ ) and 840 nm ( $\text{PA}_{840}$ ) showed a dramatic increase. Except for Cys and Hcy, other biologically relevant species including diverse amino acids, GSH,  $\text{Na}_2\text{S}$ , and other inorganic sulfur species ( $\text{HSO}_3^-$ ,  $\text{S}_2\text{O}_3^{2-}$ ,  $\text{SO}_4^{2-}$ ) showed a negligible influence on both the emission and absorption of the probe, even when a high concentration of GSH was used. Furthermore, NIR-S was applied for the *in vivo* PAI of exogenous Cys/Hcy in the subcutaneous tissue of living mice. The  $\text{PA}_{695}/\text{PA}_{840}$  value of the control group was recorded to be 0.220, while that of the Cys-treated right hind limb boosted over 10 fold was found to be 1.406. However, most of the currently reported thiol-activated probes usually exhibit responses to both Cys and Hcy. In this context, the improvement in the selectivity of PA probes towards Cys over Hcy is urgently required. Moreover, it may be realized by modulating the electrophilic nature of the response site, considering that Cys shows the highest nucleophilic activity among the three thiols (GSH, Cys, and Hcy).<sup>55</sup>

$\text{H}_2\text{S}$  ( $\text{HS}^-$ ) is a key gasotransmitter involved in signalling pathways and pathogenesis.<sup>56</sup> As another typical RS-species with a much small particle size, it could be recognized by a similar nucleophilic reaction mechanism. Shi *et al.* reported such an  $\text{H}_2\text{S}$ -sensitive and consuming nanoplatfrom (ZNNPs) for the quantitative and real-time detection of endogenous  $\text{H}_2\text{S}$ .<sup>57</sup> ZNNPs and ZNNPs@FA were obtained by encapsulation of an  $\text{H}_2\text{S}$ -responsive hydrophobic NIR-II fluorophore ZM1068-NB in amphiphilic mPEG<sub>5000</sub>-PCL<sub>3000</sub> and mPEG<sub>5000</sub>-PCL<sub>3000</sub>-FA polymers, respectively (Fig. 7a). Based on the  $\text{H}_2\text{S}$ -reactive nucleophilic addition of a 4-nitrobenzoic ester of ZM1068-NB, the ZM1068-ketone product was released, concomitant with absorption and PA spectral shifts (Fig. 7b and c). The PA signal at 680 nm ( $\text{PA}_{680}$ ) remained almost unchanged, while those at 900 nm ( $\text{PA}_{900}$ ) gradually declined against the increasing concentration of NaHS with an LOD of 0.68  $\mu\text{M}$ . Moreover, the ratiometric PA signals ( $\text{PA}_{680}/\text{PA}_{900}$ ) of ZNNPs became significantly brighter with the addition of NaHS over other interferential anions ( $\text{SCN}^-$ ,  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{NO}_2^-$ , vitamin C, GSH,  $\text{H}_2\text{O}_2$ ,  $\text{CO}_3^{2-}$ ,  $\text{S}_2\text{O}_3^{2-}$ , and  $\text{SO}_3^{2-}$ , and  $\text{AcO}^-$ ) and metal ions ( $\text{K}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Na}^+$ ,  $\text{Mg}^{2+}$ ). Next, a 2.25-fold enhancement of the  $\text{PA}_{680}/\text{PA}_{900}$  value was observed in the mice livers treated with 5 mM L-cys (a precursor for the biosynthesis of  $\text{H}_2\text{S}$ ) relative to the liver of normal mice, indicating the potential for deeply penetrated imaging of  $\text{H}_2\text{S}$ -related diseases, such as acute hepato-toxicity, brain injury, and colorectal tumors (Fig. 7d).

### 2.2.2. Reduction reaction

**Ratiometric probes.**  $\text{H}_2\text{S}$ -initiated reduction of azide is another effective mechanism to design PA probes. Xing's group developed a novel nanoprobe AzHD-LP for switch-on ratiometric PA imaging of  $\text{H}_2\text{S}$  by encapsulation of an  $\text{H}_2\text{S}$ -sensitive moiety (AzHD) in a liposome (LP).<sup>58</sup> Induced by  $\text{H}_2\text{S}$ , the electron-withdrawing azide was reduced to an electron-donating amine, accompanied by a red-shift of the absorption peak of the AzHD-LP from 600 to 700 nm (Fig. 7e). Moreover, the ratio of  $\text{PA}_{700}/\text{PA}_{532}$  increased by 23 fold with the addition of NaHS from 0 to 100  $\mu\text{M}$ . The LOD was then calculated to be 91 nM.



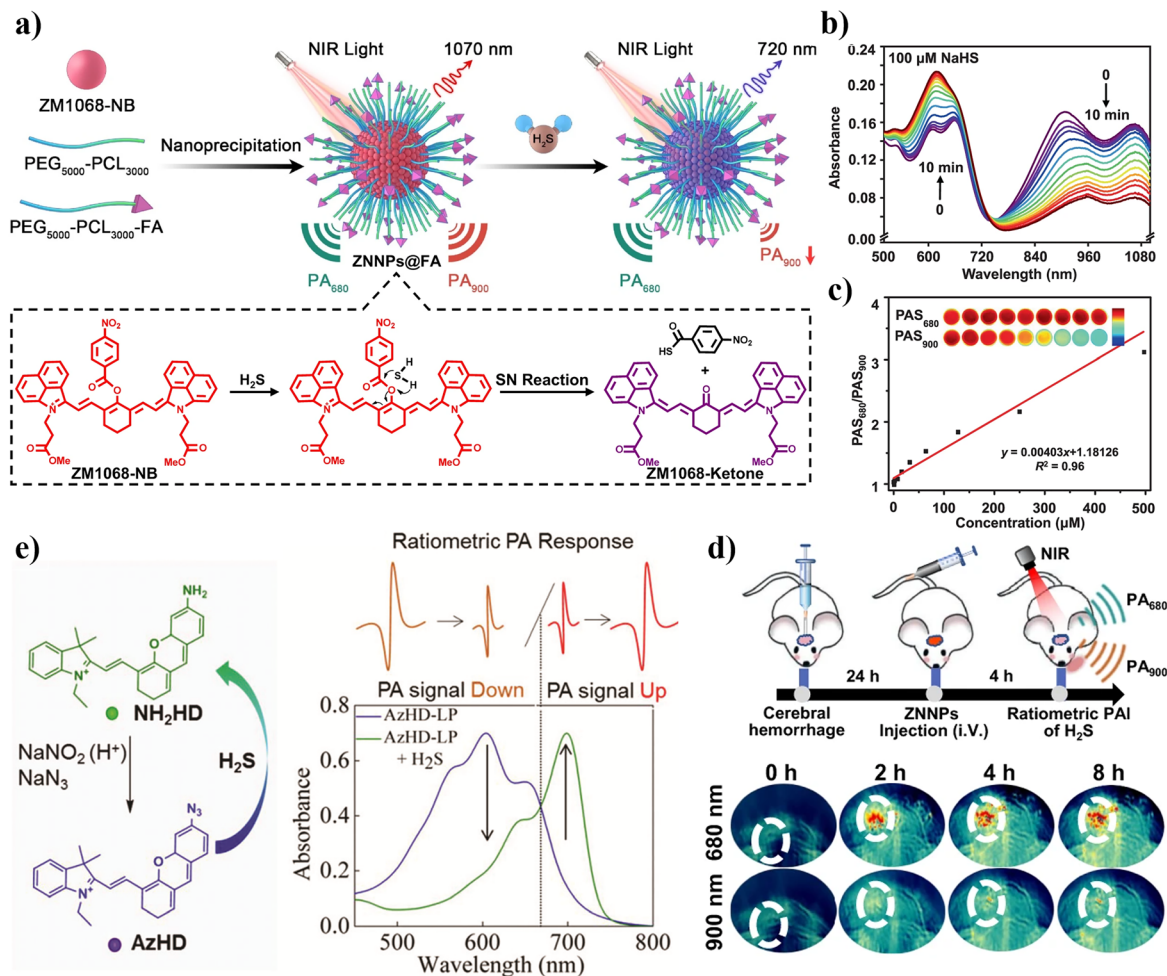


Fig. 7 (a) Schematic illustration of the fabrication of ZNNPs@FA and principle of quantitative visualization of H<sub>2</sub>S; (b) normalized absorption of ZNNPs upon incubation with NaHS; (c) linear fitting of the PAS intensity of ZNNPs with various concentrations of NaHS; (d) schematic description of intracerebral hemorrhage (ICH) model construction and PA detection of endogenous H<sub>2</sub>S by using ZNNP; (e) the proposed mechanism for the ratiometric photoacoustic detection of H<sub>2</sub>S by using AzHD. (a–d) Reprinted (adapted) with permission from ref. 57, Copyright 2022, Springer Nature; (e) Reprinted (adapted) with permission from ref. 58, Copyright 2020, Science China Press and Springer-Verlag GmbH Germany, part of Springer Nature.

Moreover, it exhibited specific responses to NaHS among various biospecies such as SO<sub>3</sub><sup>2-</sup>, CN<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, OH<sup>-</sup>, ClO<sub>3</sub><sup>-</sup>, F<sup>-</sup>, GSH and H<sub>2</sub>O<sub>2</sub>. For *in vivo* performance, the amount of H<sub>2</sub>S in brain tissues of normal and AD mice was, respectively, monitored by the ratiometric PA probe. The ratio of PA<sub>700</sub>/PA<sub>532</sub> of the normal brain tissues was 6.5-fold higher than that of the control group, while that of the AD group only displayed 1.2-fold enhancement, describing the absence of H<sub>2</sub>S in the AD mouse brain. Next, by covalent connection with a tumor targeted peptide c(RGDyK), AzHD-LP was also used as a ratiometric PA probe for intratumoral H<sub>2</sub>S monitoring in real-time in an HCT116 tumor-bearing mouse model.

Overall, RSSs exhibit distinct characteristics from RONSS. FI detection of specific RSSs has been extensively explored based on intramolecular hydrogen bonding, aromatic nucleophilic substitution, and addition reactions.<sup>59–61</sup> By contrast, PA imaging of RSSs is still not fully developed, in particular, for PAI of Hcy. Besides, PA probes for *in vivo* imaging of selenols or

thiophenols are scarce. Such PA probes may be acquired by using similar design strategies to that for imaging of GSH, Cys and Hcy, as sulfhydryl anions are the dominantly reacting constituents.

### 2.3 Imaging of other anions

Except for the reactive anion species, the PA probe can also be utilized for *in vivo* imaging of F<sup>-</sup>. Excessive F<sup>-</sup> ingestion in the human body may seriously damage the hard tissues and renal system, leading to severe diseases such as dental or skeletal fluorosis, urolithiasis, and kidney failure.<sup>62</sup> Detection of F<sup>-</sup> makes sense for precision diagnosis of diseases. On the other hand, F<sup>-</sup> is the smallest anion, and therefore features a small radius to charge ratio, indicating high reactivity towards Lewis acid. The currently reported detection mechanisms mainly include Lewis acid–base interactions, such as the high affinity of F<sup>-</sup> towards boron atoms, hydrogen-bonding interaction and breaking of the Si–O bond to alter the molecular orbital energy.

### 2.3.1. Deprotonation

**Turn-on probes.** Based on the chemical characteristics of  $F^-$ , PAI of  $F^-$  has been exemplified with a probe LET-1 reported by Lin's group.<sup>63</sup> LET-1 consisted of a naphthalimide unit as a chromophore and a phenol group as an  $F^-$ -sensitive unit. Specifically,  $F^-$ -induced deprotonation of LET-1 led to a red-shift of the light absorption from the visible 430 nm region to the NIR 640 nm region, thereby generating a PA signal at around 600 nm. The fluorescence emission at 600 nm was quenched, as the deprotonated LET-1@ exhibited a smaller highest occupied molecular orbital and the lowest unoccupied molecular orbital (HOMO–LUMO) band gap ( $E = 1.7762$  eV) than LET-1 ( $E = 3.2873$  eV). After the addition of  $F^-$  (80 mM), LET-1 displayed an over 20-fold enhancement of the PA intensity at 680 nm, with a detection limit of 0.20 mM. Besides, negligible changes were observed on the probe in response to various anions such as  $I^-$ ,  $H_2PO_4^-$ ,  $HCO_3^-$ ,  $AcO^-$ ,  $Cl^-$ ,  $NO_2^-$ ,  $CN^-$ , and  $SO_4^{2-}$ , indicating the good selectivity of LET-1 toward  $F^-$ . The *in vivo* PAI tests on mice showed a 4-fold intensity increase of  $PA_{680}$  after  $F^-$  administration, demonstrating the competence of LET-1 for *in vivo*  $F^-$  imaging.

## 3. Challenges and outlooks

The main text of the article includes activatable organic PA probe-mediated anion imaging, which is indeed significant for the *in situ* screening of pathological tissues like solid tumors as well as the visualization of physiological activities. However, the further development of PA probe-based anion imaging has still been hampered by the following three aspects: (1) the category of anion species, the *in vivo* imaging of which has been realized, remained unexpanded and the performance of the organic PA probes need to be improved; (2) quantitative detection methods are largely unexplored; and (3) instrumental limitations. Firstly, the reported organic PA probes for anion imaging are still inadequate, especially high-performance probes with excellent selectivity, sensitivity, and biocompatibility. It is also necessary to diversify the studied anion species, considering that there are much more anion species that participated in vital life processes in living systems. In addition, the accurate quantitative detection of anions is urgently required. As for the instrument, the optics (such as laser source) and electronics (such as detectors) are vital devices for PA imaging. To meet the requirements of preclinical or clinical applications, a portable and user-friendly instrument is much desirable, especially the hand-held type. However, the currently explored detectors are bulky and the laser source is much expensive, which is unfavourable for further development of PAI.<sup>64–67</sup> On the other hand, multispectral imaging is increasingly attractive for clinical application, which requires a pulse laser source with strong energy and optical parametric oscillators. The performance of the present laser source remains to be improved to fit the bill.<sup>38</sup>

Therefore, more effort could be devoted to either exploring the novel well-performed organic PA probes or modifying the existing molecules by elaborate structural design to optimize

the performance. The investigation on the excited state absorption and the triplet contribution relaxation kinetics are helpful to reinforce the photothermal conversion efficiency. Besides, combination with other materials by engineering method is beneficial for improving the blood circulation time and stability of the probes. It is necessary to establish the relationship between the property and chemical structure. The photophysical properties (photostability, strong NIR-I even II region absorption), chemical capabilities (chemical stability, sensitivity, selectivity, and water-solubility, high photothermal conversion efficiency) as well as the biological characteristics (long blood circulation, targeting ability, low toxicity, biocompatibility, biodegradability) and economic cost should be comprehensively considered. For instance, the PA probes with NIR-II absorption and high absorption coefficients are expected to achieve high spatial resolution and excellent imaging contrast. Generally, the strong NIR absorption of PA probes is related to their large structural conjugation, which however may lead to high molecular rigidity, and consequently poor biocompatibility. In addition to the photophysical properties, the biological characteristics are also essential considerations as most of the organic PA probes are limited by their poor water solubility and rapid metabolism from the living body. Introducing the hydrophilic moieties into the skeleton to increase the solubility may inversely cause a reduced absorption intensity and hypochromic shift. Nano-engineered probes are usually excreted by the reticuloendothelial system. However, it increases the risk of organ toxicity due to excess enrichment of nanoprobe in the liver and spleen.<sup>68</sup> To reduce the safety concerns, renal-clearable PA probes have been increasingly explored, due to the minimal metabolism process and the accessibility to anion imaging of kidneys.<sup>69</sup> Furthermore, new detection methods or responsive mechanisms for optimal *in vivo* anion imaging are required. Quantitative detection and reversible imaging, which is necessary for dynamic monitoring, are lacking. Also, the challenges of the instrumental limitations of poor resolution, narrow dynamic range, and confined laser wavelength need to be addressed to better match the characteristics of probes and realize high resolution and contrast imaging outcomes. The technic parameters of both optics (*e.g.*, cost, pulse length and energy of laser source) and electronics (*e.g.*, sensitivity, size and array conformation of detectors) need to be optimized to promote the generalization of PAI.

## 4. Conclusions

The recent advances of such probes are summarized in this review, with a particular emphasis on specific anion-induced activation mechanisms of these PA probes at the molecular level. The anion-recognition of these PA probes involves diverse chemical reactions, mainly including cleavage, spiro lactone ring-opening, oxidation, degradation, nucleophilic substitution, and reduction reactions, and the deprotonation process. On the basis of these mechanisms, the rational structural

design of the analyte-response unit of the activatable PA probes could be achieved to develop anion-responsive PA probes with excellent selectivity and sensitivity. Finally, the challenges and perspectives are also discussed. Much effort should be devoted to (1) developing high-performance anion-responsive organic PA probes by nanoengineering and/or the chemical modification approach, (2) innovating detection methods and (3) combining PAI with other imaging technologies to achieve multispectral, multimodal and multifunctional detection of anion species *in vivo*. In this way, activatable organic PA probes for *in vivo* imaging of bio-functional anion species, as an important part of the *in vivo* molecular imaging platform, hold great promise for unveiling the involved physiological and pathological secrets.

## Abbreviation

AcO <sup>-</sup>	Acetate ion
Arg	Arginine
CT	Computed tomography
Cys	Cysteine
cRGD	Tumor-targeting short peptide c(RGDfK)
DSPE-PEG <sub>2000</sub>	1,2-Distearoyl- <i>sn</i> -glycero-3-phosphoethanolamine- <i>N</i> -[methoxy(polyethylene glycol)-2000]
DSPE-PEG <sub>2000</sub> <sup>-</sup>	
COOH	1,2-Distearoyl- <i>sn</i> -glycero-3-phosphoethanolamine- <i>N</i> -[carboxy(polyethylene glycol)-2000]
F <sup>-</sup>	Fluoride ion
FI	Fluorescence imaging
GSH	Glutathione
HClO/CLO <sup>-</sup>	Hypochlorite
HOMO	Highest occupied molecular orbital
H <sub>2</sub> O <sub>2</sub>	Hydrogen peroxide
Hcy	Homocysteine
LUMO	Lowest unoccupied molecular orbital
LP	Liposome
MRI	Magnetic resonance imaging
NIR	Near-infrared
NIR-II	The second near-infrared window
NIRF	Near-infrared fluorescence imaging
<sup>1</sup> O <sub>2</sub>	Singlet oxygen
ONc	5,9,14,18,23,27,32,36-Octabutoxy-2,3-naphthalocyanine
ONOO <sup>-</sup>	Peroxynitrite
O <sub>2</sub> <sup>•-</sup>	Superoxide anion
PAI	Photoacoustic imaging
PA	Photoacoustic
PEG	Polyethylene glycol
PEG- <i>b</i> -PPG- <i>b</i> -PEG	Poly(ethylene glycol)- <i>b</i> -poly(propylene glycol)- <i>b</i> -poly(ethylene glycol)
PET	Positron emission tomography
RONS	Reactive oxygen and nitrogen species
RSS	Reactive sulfur species

RS <sup>-</sup>	Thiolates
S <sub>1</sub>	First excited state
S/N	Signal-to-noise ratio
TPB	Triphenyl borane
US	Ultrasound imaging
<i>i.v.</i>	Intravenous injection
<i>s.c.</i>	Subcutaneous injection
<i>oral</i>	Oral administration

## Conflicts of interest

There are no conflicts to declare.

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