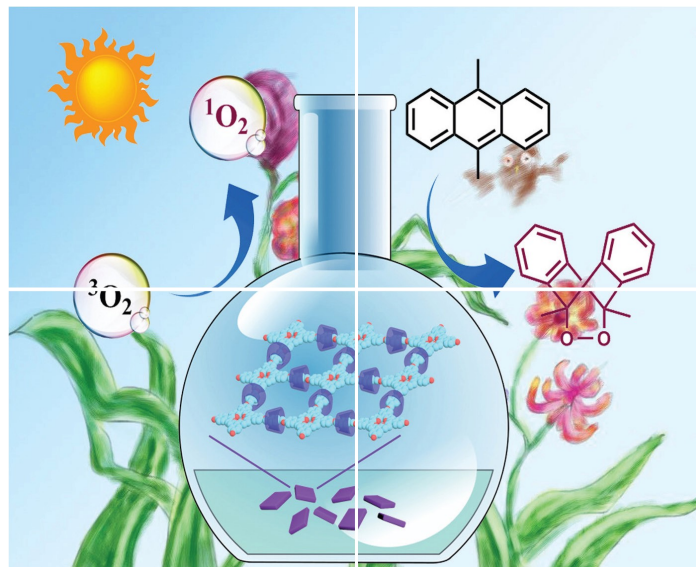


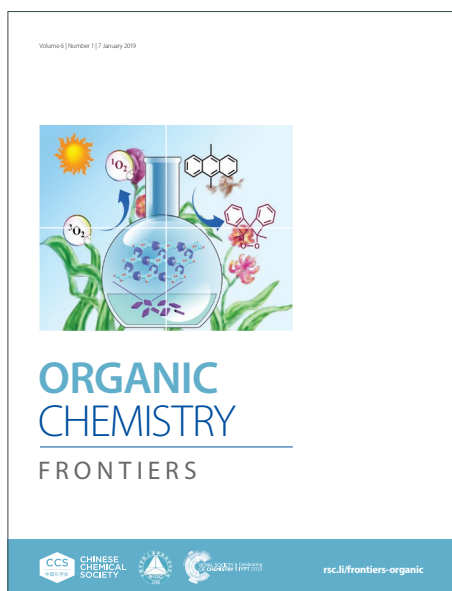
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Ru(II)-Catalyzed Direct Alkynylation of 2-Acylimidazoles with TIPS-Alkynes or TIPS-Bromoalkynes under Air

Received 00th January 20xx,
Accepted 00th January 20xxYuting Gui,^{†a} Xingchi Li,^{†a} Ting Fu,^a Taoyuan Liang,^a Zequan Li,^b Shuangliang Zhao^{*,a} and Zhuan Zhang^{*,a}

DOI: 10.1039/x0xx00000x

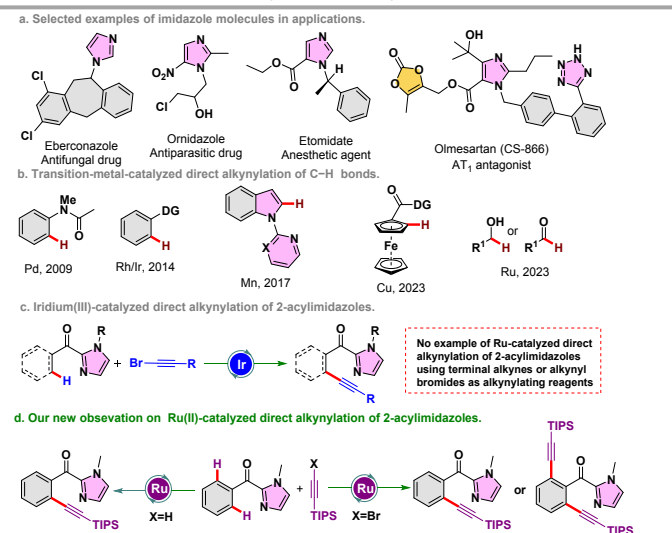
The Ru(II)-catalyzed *ortho*-alkynylation of 2-acylimidazoles is reported, demonstrating compatibility with TIPS-protected terminal alkynes or TIPS-protected bromoalkynes as alkynylating reagents. The reaction tolerates a wide range of functional groups, and the resulting alkynylated products can be readily transformed into high-value compounds, presenting promising applications in medicinal chemistry and materials science. This strategy addresses the existing gap by proposing that the alkynylation reaction proceeds through six-membered ruthenacycle intermediates.

Introduction

Imidazole and its derivatives are highly significant due to their crucial roles in medicinal chemistry and materials science.¹ Their remarkable versatility has led to their incorporation into numerous clinically used drugs, including antifungal, antiparasitic, anesthetic, and antihypertensive agents, among others, which exhibit high therapeutic efficacy and market value (Scheme 1a).² Furthermore, 2-acylimidazoles serve as pivotal molecular building blocks, enabling the synthesis of a wide array of bioactive compounds.³ Their adaptability in structural modification underscores their potential as powerful tools in organic synthesis, offering promising avenues for developing novel pharmaceuticals and functional materials.⁴

Alkynes are fundamental building blocks in synthetic chemistry and materials science.⁵ While aryl alkynes are commonly synthesized from aryl halides *via* the Sonogashira reaction,⁶ a more desirable and attractive approach leverages the abundance of C–H bonds in arenes through transition-metal-catalyzed C–H activation. This strategy has been extensively explored over the past several decades, leading to the development of numerous synthetic methods for constructing complex structures.⁷ Transition-metal-catalyzed direct alkynylation of C–H bonds is particularly advantageous, as it enables direct functionalization without the need for preactivation—a key benefit in the late-stage modification of complex molecules. Pioneered by the work of Chatani,⁸ Li,⁹ Ackermann,¹⁰ Yu,¹¹ and Krische,¹² direct alkynylation of C–H bonds has been achieved through the exploration of diverse

transition-metal catalysts, enabling access to a wide array of molecular architectures (Scheme 1b).



Scheme 1 Previous works and our new observation

Extensive efforts have been devoted to the functionalization of 2-acylimidazoles due to their broad applications. Various transformations, including arylation, alkylation, alkenylation, amidation, esterification, and trichloromethylation, have been explored.^{3b, 13} In 2020, Chatani and coworkers reported an Ir-catalyzed alkynylation of 2-acylimidazoles using bromoalkynes, leveraging the imidazole moiety as a directing group through a novel chelation system (Scheme 1c).¹⁴ Despite these advances, Ru-catalyzed direct alkynylation of 2-acylimidazoles remains unexplored. Herein, we disclose a Ru(II)-catalyzed protocol for the direct alkynylation of 2-acylimidazoles with terminal alkynes or alkynyl bromides as versatile alkynylating reagents *via* six-membered ruthenacycle intermediates for the first time. This method features different alkynylating reagents available, high atom economy, excellent functional group tolerance, and the unique advantage of

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[†] Electronic supplementary information (ESI) available: Experimental details and NMR spectra for all compounds. CCDC 2417911 (4sa). For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/x0xx00000x

[‡]These authors contributed equally to this work.

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facile imidazole removal, enabling straightforward conversion to esters under mild conditions (Scheme 1d).

Results and discussion

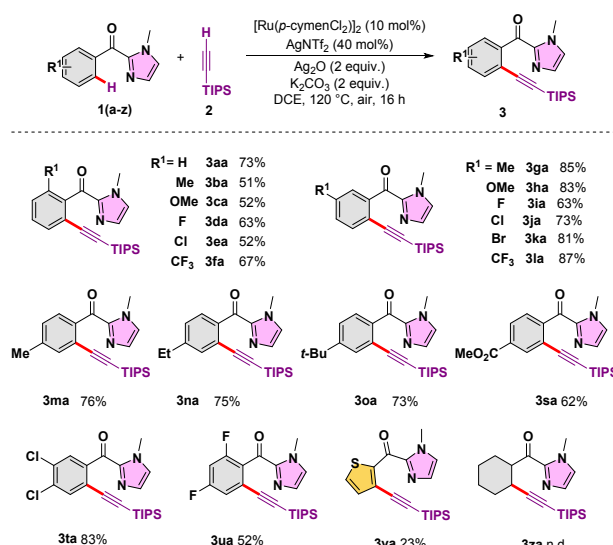
Table 1 Optimization of the reaction conditions^a

Entry	Variation from "standard conditions"	Yield (%) ^b 3aa/4aa
1	none	73/–
2	CH ₃ CN instead of DCE	64/–
3	toluene instead of DCE	31/–
4	1,4-dioxane instead of DCE	50/–
5	AgOAc instead of Ag ₂ O	63/–
6	Ag ₂ CO ₃ instead of Ag ₂ O	59/–
7	Cu(OAc) ₂ instead of Ag ₂ O	n.d/–
8	K ₃ PO ₄ instead of K ₂ CO ₃	23/–
9	NaOAc instead of K ₂ CO ₃	31/–
10	Cs ₂ CO ₃ instead of K ₂ CO ₃	29/–
11	without [Ru(<i>p</i> -cymene)Cl ₂] ₂	n.d/–
12	without AgNTf ₂	31/–
13	without K ₂ CO ₃	n.d/–
14	without Ag ₂ O	13/–
15	none	67/21
16	without K ₂ CO ₃	n.d/0
17	without AgNTf ₂	11/0
18	without [Ru(<i>p</i> -cymene)Cl ₂] ₂	n.d/0
19	CH ₃ CN instead of DCE	21/0
20	HFIP instead of DCE	51/21
21	toluene instead of DCE	23/0
22	1,4-dioxane instead of DCE	51/25
23	K ₃ PO ₄ instead of K ₂ CO ₃	36/5
24	NaOAc instead of K ₂ CO ₃	41/7
25	Na ₂ CO ₃ instead of K ₂ CO ₃	33/21
26	3.0 equiv. K ₂ CO ₃ for 36 h	16/73

^aReaction conditions for **3aa**: **1a** (0.20 mmol), **2a** or **2a'** (entries 1–25, 0.30 mmol), [Ru(*p*-cymene)Cl₂]₂ (10 mol%), AgNTf₂ (40 mol%), Ag₂O (entries 1–14, 2 equiv.), K₂CO₃ (2 equiv.) in DCE (1.5 mL) at 120 °C for 16 h. ^aReaction conditions for **4aa**: **1a** (0.20 mmol), **2a'** (entry 26, 0.44 mmol), [Ru(*p*-cymene)Cl₂]₂ (10 mol%), AgNTf₂ (40 mol%), K₂CO₃ (3 equiv.) in DCE (1.5 mL) at 120 °C for 36 h. ^bIsolated yield.

The optimization of the alkylation of (1-methyl-1*H*-imidazol-2-yl)benzophenone **1a** was investigated using either TIPS-protected terminal alkyne **2a** or TIPS-protected bromoalkyne **2a'** as the alkynyl source (Table 1). We initially focused on oxidative *ortho* alkylation. Employing 10 mol% [Ru(*p*-cymene)Cl₂]₂ as the catalyst in DCE, along with AgNTf₂ (40 mol%), Ag₂O (2 equiv.), and K₂CO₃ (2 equiv.), afforded the

desired alkylation product **3aa** in 73% yield (entry 1). However, replacing DCE with acetonitrile, toluene, or 1,4-dioxane resulted in lower yields (entries 2–4). We then screened various additives, including AgOAc, Ag₂CO₃, Cu(OAc)₂, K₃PO₄, NaOAc, and Cs₂CO₃, but none provided improved results (entries 5–10). Control experiments confirmed that [Ru(*p*-cymene)Cl₂]₂, AgNTf₂, Ag₂O, and K₂CO₃ were essential for the reaction (entries 11–14). Next, we optimized the reaction conditions for *ortho* alkylation using **2a'** as the alkynyl source. To our delight, employing 10 mol% [Ru(*p*-cymene)Cl₂]₂, along with AgNTf₂ (40 mol%) and K₂CO₃ (2 equiv.) in DCE, yielded the alkylation product **3aa** in 67% yield with an acceptable selectivity (entry 15). Control experiments further confirmed that [Ru(*p*-cymene)Cl₂]₂, AgNTf₂, and K₂CO₃ were indispensable (entries 16–18). Solvent screening revealed that alternative solvents resulted in diminished yields (entries 19–22), and testing different bases did not lead to improved outcomes (entries 23–25). Finally, we discovered that increasing K₂CO₃ to 3 equiv. and extending the reaction time to 36 h led to the formation of the dialkylated product **4aa** in 73% yield (entries 26). Notably, these optimized reaction conditions are air-stable and can be conveniently carried out on the benchtop.

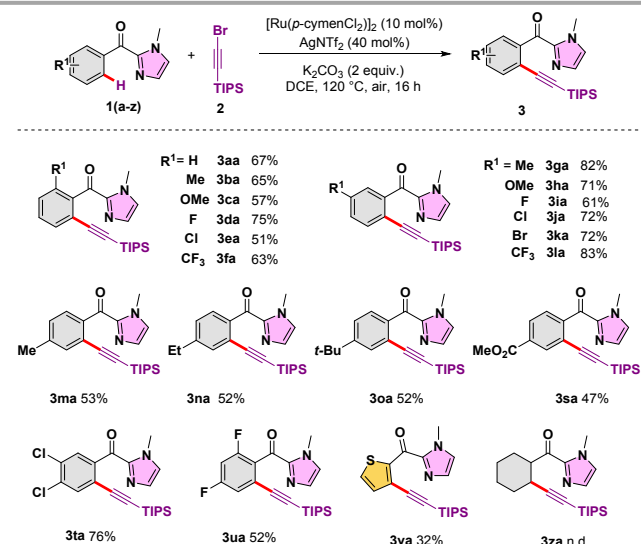
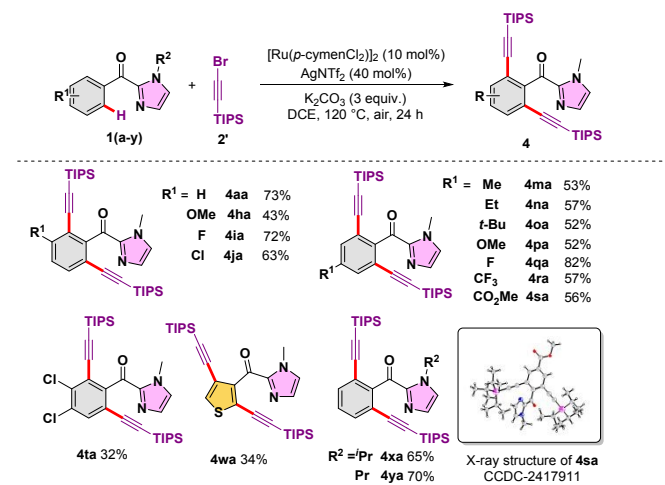


Scheme 2 Scope of oxidative *ortho*-alkynylation

Next, we explored the scope of the oxidative *ortho*-alkynylation reaction under standard conditions (Scheme 2). A variety of electron-withdrawing and electron-donating groups at the *ortho*, *meta*, and *para* positions were well tolerated, affording the desired alkylation products (**3aa–3sa**) in yields ranging from 51% to 87%. Notably, substrates bearing double electron-withdrawing groups led to the target products (**3ta** and **3ua**) in good to excellent yields. Additionally, replacing the phenyl group with a thiophene ring was compatible with this strategy, yielding **3va** in 23% yield. However, cyclohexyl-substituted 2-acylimidazole failed to produce the expected product **3za**.

Encouraged by the preliminary results, we next explored the scope of *ortho*-monoalkynylation under the optimized

conditions (Scheme 3). A variety of substituents on the benzene ring, including methyl, ethyl, *tert*-butyl, methoxy, fluorine, chlorine, bromine, trifluoromethyl, and methyl carbonate, were well tolerated, affording the desired products (**3aa–3sa**) in yields ranging from 47% to 83%. Disubstituted substrates successfully generated **3ta** and **3ua** with yields of 76% and 52%, respectively. Interestingly, the substrate containing a thiophene ring produced the corresponding product **3va** in 32% yield. Unfortunately, the substrate **1z** failed to furnish the expected product **3za**.

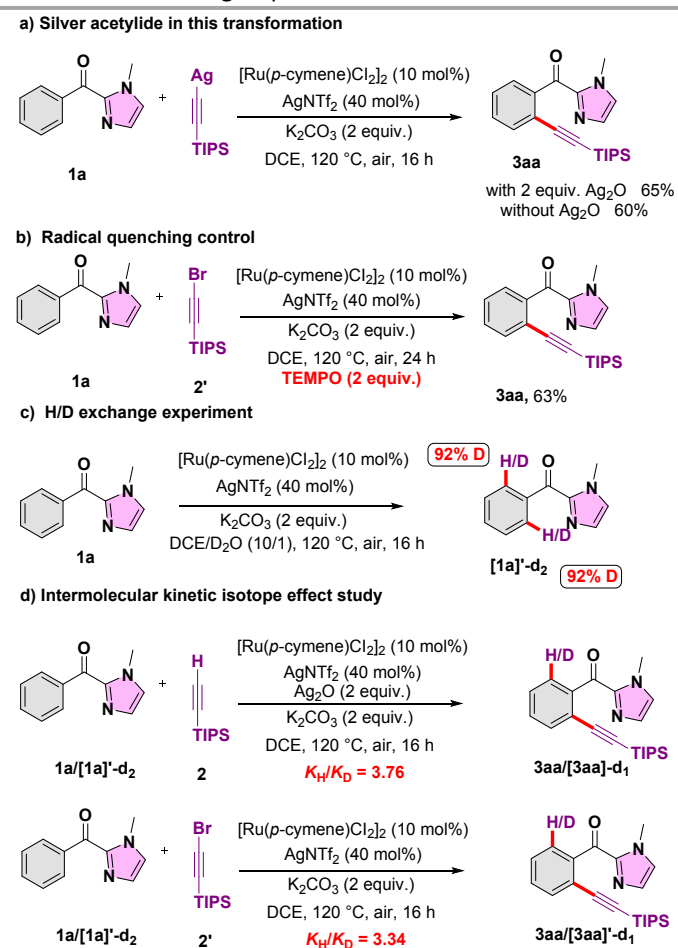
Scheme 3 Scope of *ortho*-monoalkynylationScheme 4 Scope of *ortho*-dialkynylation

Furthermore, we evaluated the reaction scope for *ortho*-dialkynylation (Scheme 4). Various 2-acylimidazole derivatives substituted at the *meta* and *para* positions on the benzene ring were well tolerated, affording the desired products (**4aa–4sa**) in good to excellent yields. The structure of the product **4sa** was confirmed by X-ray diffraction analysis. Delightfully, the substrate **1t** could afford the corresponding product **4ta** in a relatively lower yield due to the steric hindrance. Encouragingly, the optimized reaction conditions were also effective for the formation of the thienyl product **4wa** in 34% yield. Additionally,

different *N*-substituents were well tolerated, yielding **4xa** and **4ya** in 65% and 70%, respectively.

DOI: 10.1039/D5QO00868A

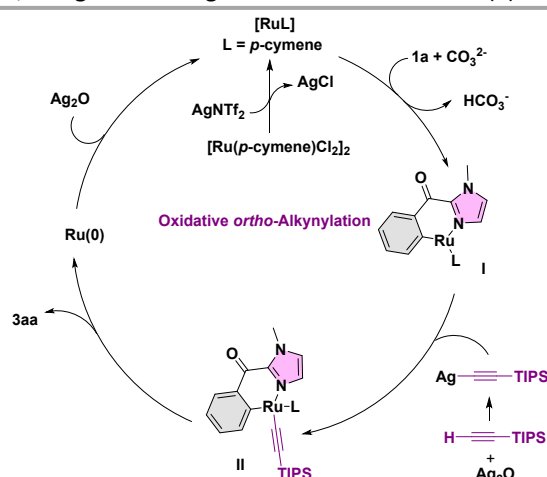
As the detailed mechanism remains to be elucidated, a series of preliminary mechanistic studies were conducted (Scheme 5). First, silver acetylide was successfully synthesized and subjected to oxidative *ortho*-alkynylation with Ag₂O or without Ag₂O under standard conditions, affording the desired product **3aa** in 65% and 60%, respectively (Scheme 5a). This result suggests that the reaction involves the formation of an alkynyl-Ag species, which may undergo transmetalation to generate a Ru-alkynyl intermediate. Next, the addition of the radical scavenger TEMPO had no effect on the *ortho*-alkynylation process, indicating that a radical pathway is not involved (Scheme 5b). Furthermore, H/D scrambling experiments conducted in the absence of compound **2a** demonstrated the reversibility of C–H bond cleavage (Scheme 5c). Finally, the kinetic isotope effect (KIE) values of *K_H*/*K_D* were determined to be 3.76 for the oxidative *ortho*-alkynylation and 3.34 for the *ortho*-alkynylation (Scheme 5d). These results suggest that C–H bond cleavage is the rate-determining step in both reactions.



Scheme 5 Control experiments

Based on the aforementioned studies and previous research, we propose two possible catalytic mechanisms (Scheme 6). In the oxidative *ortho*-alkynylation using a TIPS-protected terminal alkyne **2a** as the alkynyl source,¹⁵ the Ru(II) catalytic precursor is initially generated in the presence of AgNTf₂. This species

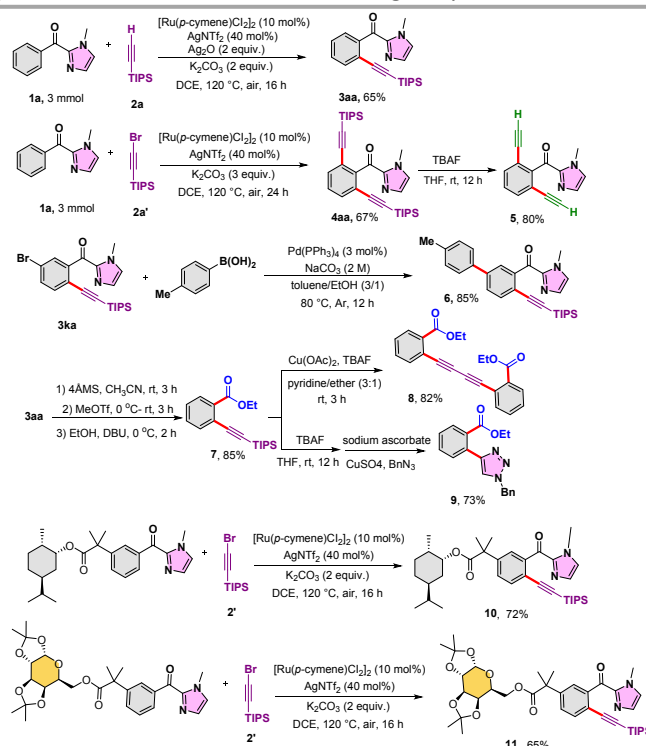
coordinates with substrate **1a** and undergoes base-assisted deprotonation at the *ortho*-C–H bond to form the ruthenacycle intermediate **I**. Intermediate **I** then undergoes transmetalation with the in situ generated alkynyl silver species to afford the alkynyl–ruthenium complex **II**. Subsequent reductive elimination furnishes the product **3aa** and a Ru(0) species, which is reoxidized by Ag₂O to regenerate the active Ru(II) catalyst. In the *ortho*-alkynylation using a TIPS-protected bromoalkyne **2a'** as the alkynyl source,¹⁶ a similar six-membered ruthenacycle intermediate **I** is formed and coordinates with substrate **2a'**. This is followed by alkyne incorporation to generate intermediate **IV**, which undergoes base-assisted β -bromide elimination to deliver the product **3aa** and KBr, along with the regeneration of the active Ru(II) catalyst.



Scheme 6 Plausible reaction pathways

To evaluate the practical applicability of this strategy, gram-scale reactions and post-synthetic modifications were carried out (Scheme 7). The products **3aa** and **4aa** were obtained in 65% and 67% yields, respectively, on a 3.0 mmol scale. Subsequent deprotection of the TIPS group using TBAF afforded the terminal di-alkyne product **5** in 80% yield, representing a more synthetically valuable compound. Furthermore, the bromo-substituted product **3ka** underwent a Suzuki coupling reaction with phenylboronic acid to afford the corresponding product **6** in 85% yield. Notably, the imidazole group could be readily removed to generate ester **7**, which was further transformed into valuable derivatives **8** and **9**, demonstrating the broad

synthetic utility of this protocol. Finally, under standard conditions with a TIPS-protected bromoalkyne as the alkynyl source, 2-acylimidazole derivatives containing drug-like molecular scaffolds were well tolerated, and the target products **10** and **11** were obtained in good yields.



Scheme 7 Gram-scale synthesis and transformations

Conclusions

In summary, we have developed a Ru(II)-catalyzed *ortho*-alkynylation of 2-acylimidazole derivatives using either TIPS-protected terminal alkynes or TIPS-protected bromoalkynes. This protocol offers a valuable approach for the synthesis of diverse molecular building blocks and is particularly promising for late-stage functionalization, with potential applications in medicinal chemistry and materials science.

Author contributions

Y. Gui, X. Li and T. Fu performed the experiments, collected and analyzed the data. Y. Gui and X. Li wrote the original draft. T. Liang and Z. Li assisted in revising the manuscript. S. Zhao and Z. Zhang conceived and directed the project and revised the manuscript. All authors discussed the results and commented on the manuscript.

Data availability

The data supporting this article have been included as part of the ESI.†

Conflicts of interest

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There are no conflicts to declare.

Acknowledgements

This work is supported by the National Natural Science Foundation of China (22461003), the Natural Science Foundation of Guangxi Province (2025GXNSFAA069547), Guangxi Major Talent Project, and the Opening Project of Guangxi Key Laboratory of Petrochemical Resource Processing and Process Intensification Technology (2023K017).

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Data Availability Statement

The data underlying this study are available in the published article and its ESI. The Supporting Information is available free of charge on the RSC Publications website at DOI:
Detailed experimental procedures, characterization data, crystallographic data for **4sa** (CIF), copies of NMR spectra for all isolated compounds.

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